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Distinction between
elements and
compounds; properties
of each; same elements
may form different
compounds • Reading
chemical formulas:
compound names,
element names, number

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and ratio of atoms •

Chemical bonds between atoms involve electrons • Ionic Bonds: result from electron transfer; reading superscript of ions; examples of ionic bonds; types of elements forming ...

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Bonding

Answers

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Chemical Bonding.

SECTION 1. SHORT

ANSWER Answer the
following questions in
the space provided. 1.

aA chemical bond
between atoms results
from the attraction
between the valence

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electrons and of
different atoms.

(a)nuclei(c)isotopes.

(b)inner

electrons(d)Lewis

structures. 2.b A

covalent bond consists
of.

6 Chemical Bonding

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CHAPTER 6 REVIEW

Chemical Bonding

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SECTION 1 SHORT

ANSWER Answer the following questions in the space provided. 1. a

A chemical bond between atoms results from the attraction between the valence electrons and of different atoms. (a) nuclei (c) isotopes (b) inner electrons (d) Lewis

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1. The valency of an element is _____ (a) the combining capacity of one atom of it (b) the number of bonds

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formed by its one atom
(c) the number of
hydrogen atoms that
combine with one atom
of it (d) all the above
Answer. (d)

Multiple Choice
Questions On Chemical
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if the difference of two
atoms'
electronegativities is

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higher than 1.7, it is ionic; if it is more than 0.3 but less than 1.7, it is polar covalent; if it is 0.3 or less, it is non-polar covalent. (type of bond) H and F. $4.0 \text{ (F)} - 2.1 \text{ (H)} = 1.9$; 1.9 is higher than 1.7 so ionic. (type of bond) Cu and S.

6.1 Introduction to
Chemical Bonding
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Chemical bonds? In ionic bonding, many atoms transfer electrons. The atoms are then held together by the attraction of opposite charges. In covalent bonding, atoms share electrons and form independent molecules.

Types of Chemical Bonding A chemical bond is a mutual electrical attraction

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between the nuclei and
valence electrons of
different atoms that

Answers

CHAPTER 6 Chemical
Bonding

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Chemical Bonding and
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13 Chemical Bonding

CHAPTER How Does
Chemical Bonding Help
to Explain the

Properties of Chemical
Substances? Rocks, coal,
soil, petroleum, trees,
and the human body are
all complex mix-tures of
chemical compounds in
diff erent
compositions. Chemical
compounds are made
up of 2 or more

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elements that are bonded together.

Although there are substances in nature that consist of unbonded atoms, they are very rare.

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