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MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question.

1) Consider the following reaction:  $3A \rightarrow 2B$  The average rate of appearance of B is given by  $D[B]/Dt$ . Comparing the rate of appearance of B and the rate of

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Test 1. Which of the following does NOT influence the  
speed of a chemical reaction? a) concentration of  
reactants b) nature of reactants c) temperature d)  
presence of a catalyst e) none of these 2. What would  
cause the change in the kinetic energy diagrams as  
shown? KE KE a) increasing the H

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Chemistry 12 Kinetics Practice Test # 2 1. Which of the following units could be used to express the reaction rate? A. mL/s B. mL/g C. g/mL D. mL/mol 2.

Consider the reaction:  $\text{Zn (s)} + 2\text{HCl (aq)} \rightarrow \text{ZnCl}_2\text{(aq)} + \text{H}_2\text{(g)}$  The rate of production of  $\text{ZnCl}_2$ , can be increased by A. decreasing the  $[\text{HCl}]$ . B. increasing the temperature

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Ib Topic 6 & 16 Chemical Kinetics Revision: - Lessons  
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Questions ...

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NEET Catalyst: Practice Questions of Chemical Kinetics ...

Kinetics Practice Exam Note: All problems included in this practice exam are drawn from problems used in previous semesters. Exams typically include 7 or 8 problems that are a mixture of qualitative problems calling for written explanations and quantitative problems that involve calculations and, in some cases,

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written explanations.

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