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ANSWER: Correct Chapter 9 Reading Question 2 Part A Iron reacts with oxygen in the air to produce rust (Fe2O3) according to the reaction shown below. 4 Fe + 3 O2 2 Fe2O3 How many moles of rust can be formed if 1.8 mols of iron are present? Hint 1. Begin with your given number of moles.

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**Mastering Chemistry Chapter 8 Answer Key**

Mastering Chemistry Chapter 2. Rank from most to fewest valence electrons of the following elements: Kr, Ga, Br, As, K, - Answer: Kr (8), Br (7), As (5), Ga (3), K (1). Valence electrons are equal to the atom's main group number. Link: <https://www.coursehero.com/file/20333255/Mastering-Chemistry-Chapter-2/>.

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